- 1. SAKLINSKIY, V. V.
- 2. USSR (600)
- 4. Automobiles Apparatus and Supplies
- 7. Examples of the application of metallo-ceramic automobile and tractor parts.

  Avt. trakt. promo. no. 11, 1952

9. Monthly List of Russian Accessions, Library of Congress, March 1953, Unclassified.

SAKL	. INSKIY,	v. v.	, ENG.							
3 1	ls, Pow									
Bime	etallic	metal-	ceramic	parts.	Vest.mash.	32 No. 3,	1952.			
Mon	thly Li	st of 1	Russian	Accessio	ons, Library	y of Congr	ess, Octo	ber 1952	2. Uncla	ssified.

RAKOWSKIY, V.S.; SAKLINSKIY, V.V.; 5030L0VA, T.F., tekhnicheskiy redaktor;

[Powdered metals in machine building; a reference nanual] Metallokeramika v meshinostroenii; spravochnos posobie. Moskva, Gos. nauchnotekhn. izd-vo mashinostroit. lit-ry, 1055. 71 p.

(Powder metallurgy)

(Powder metallurgy)

SAKLINSKIY.

Cetegory: USSR/Solid State Physics - Diffusion. Sintering E-6

Abs Jour : Ref Zhur - Fizika, No 3, 1957, No 6691

: Seklinskiy, V.V. Author

: Niitavtoprom (Scientific Research Institute of Automotive Inst

Industry), USSR

: Investigation of the Sintering Frocess of Certain Netel Title

Fowders.

Orig Fub : Poroshkovaya matellurgiya, Yeroslavl', 1955, 21-51

Abstract: During the process of sintering of metal powders a contact is formed and made closer between particles of the powders as a result of diffusion processes, caused by the temperature mobility of the atoms. The surface diffusion predominates during the initial stage, while volume diffusion predominates in the final stage. The process of formation of the contact is attributed to forces causing plastic diffusion of the powder perticles, end to the action of surface tension that results from liquidation of defects of the

crystalline lattice.

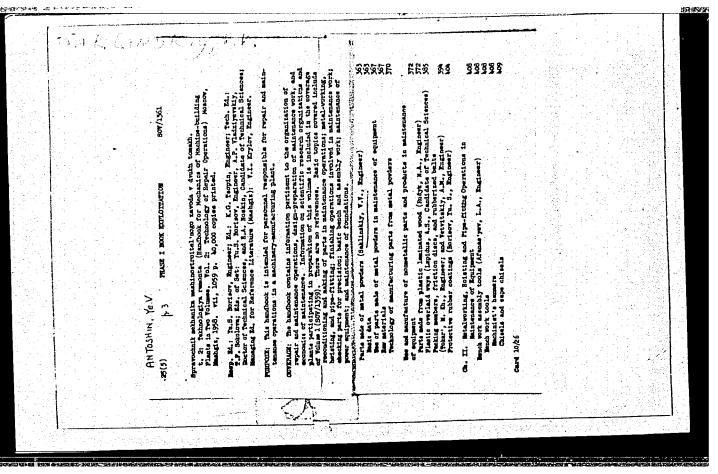
: 1/1 Card

SAKLINSKIY, V. V.	
"Gesinterte Bimetall Lage	
paper presented at Intl.	Powder etallurgy Meeting , Eisenach, 28-31 May 1957.
sie Tecanik, No. 10, Oct	
	일반 : 보고 기존 현실 보는 기존에 가능한 등로 보고 함께는 그를 하는 것은 기존에 가는 하는데 기존 기존 기존 보고 있는데 기존
	는 경험을 보고 있다. 현실 경기 등에 되었다. 그는 경기가 되었다. 그런 그는 그는 그를 모르는 것을 받는 것이 되었다. 그는 것이 되었다. 그는 것이 되었다. 그는
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[Handbook for machinists of machinery plants in two volumes] Soravochnik mekhanika mashinostroitel'nogo zavoda v dvukh tomakh. Moskva, Gos. nauchno-tekhn. izd-vo mashinostroit. lit-ry. Vol. 2.[The technology of repair work] Tekhnologiia remonta. Otv. red. toma IU. S. Borisov. (MIRA 11:10)

(Machinery--Maintenance and repair)
(Machine-shop practice)



AUTHOR: Yevseyev, A.S., Saklinskiv, Valvantana

113-58-7-14/25

TITLE:

Powder Metallurgy in the Automobile Industry (Poroshkovaya

metallurgiya v avtomobil'noy promyshlennosti)

PERIODICAL:

Avtomobil naya promyshlennost, 1958, Nr 7, pp 29-30 (USSR)

ABSTRACT:

In Soviet automobile production powder metallurgy assumes an ever growing industrial importance. From 1954 to 1957, the output of metallo-ceramic parts in automobile plants increased from 2.5 to 115 tons. This figure will increase by several times in 1958 and by 15 times by 1965. The high accuracy and antifriction properties of the metallo-ceramic parts have been used for the bushings of the guiding valves of the booster engines of the "Volga" and "Moskvich" cars and parts of the gearbox of the ZIL-111 and "Volga" cars. The Yaroslavskiy and Gor'kovskiy avtozavody (Yaroslavl' and Gor'kiy Automobile Plants) and several other car plants are already using powder metallurgy in large amounts. The characteristics of various metal powders are given (Table 1), as well as a brief characteristic of the necessary equipment. NIITAvtoprom, in cooperation with other plants, has done very much in introducing powder metallurgy to the automobile industry. Automobile types and the parts produced with powder metallurgy base are indicated

Card 1/2

Powder Metallurgy in the Automobile Industry 113-58-7-14/25
in table 2. There is 1 photo and 2 tables.

ASSOCIATION: NIITAvtoprom (NIITAvtoprom)

1. Powder alloys—Applications 2. Automobiles—Production

Card 2/2

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		(Powder metal processes)	
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	化复数粉 化二氯甲烷酸异丙基	لأباعث ومعريد تبريت والحاجات والحائل ولاتحاك وكالمركب الخاجا المرجوا ألهران	시시 회사 기사 사는 사람들이 가는 물건이 하는

SAKLINSKIY, V.V.; BLAGIN, V.I.

Present state and the outlook for the development of powder metallurgy in the automobile industry. Avt. prom. 30 no.6: 29-31 Je '64.

1. Nauchno-issledovatel'skiy institut tekhnologii avtomobil'noy promyshlennosti i Gor'kovskiy avtomobil'nyy zavod.

Development and use of new materials and processes of the powder metallurgy in the automobile industry. Avt.prom. 31 no.7:33-35 Jl '65. (MIRA 18:8)
1. Nauchno-issledovatel'skiy tekhnologicheskiy institut avtomobil'noy promyshlennosti.

5/122/61/000/006/008/011 D244/D301

AUTHOR:

Saklinskiy, V.V., Engineer

TITLE:

Engineering and economic effectiveness of powder

metallurgy

PERIODICAL: Vestnik mashinostroyeniya, no. 6, 1961, 52-54

TEXT: Methods of powder metallurgy are highly economical and effectively supply the increasing demands of the automobile industry for some small components. The following comparison shows the growth of production by these methods:

Typical applications of powder metallurgy are:

For antifriction components to carry high loads at high speeds with inadequate lubrication; for friction materials to ensure a coefficient of friction of at least 0.1; as porous materials for fuel filters; as substitutes to economize on non-ferrous metals. The production

Card 1/3

APPROVED FOR RELEASE: 09/19/2001 CIA-RDP86-00513R001446810014-2"

Engineering and economic...

S/122/61/000/006/008/011 D244/D301

technology consists of mixing the powders and sintering the mixtures at pressure to form the required component which is then checked for accuracy, sulphided to reduce friction and annealed to improve the structure and machining properties. For friction components only sintered metal ceramic powders give a friction coefficient in oil higher than 0.1. These are used in automobiles 311-111 (Z1L-111) Chayka, Volga etc. The use of sintered components can result in large savings. The Kaluzhskiy zavod avtomotoelektro-oborudovaniya (Kaluga Plant of Automobile Electrical Equipment) replaced alloy bearings of electric motors by bearings in iron metal ceramic to save 3000 rubles per ton. On the average 1 ton of automobile components required 1220 man hours if produced by ordinary methods: but only 1064 man hours by the techniques of powder metallurgy. Due to improvements in technology and the increased volume, the productivity of the industry has greatly increased (Table 2) Further development of the industry is envisaged in the present Seven Year Plan to increase the output by more than 30 times.

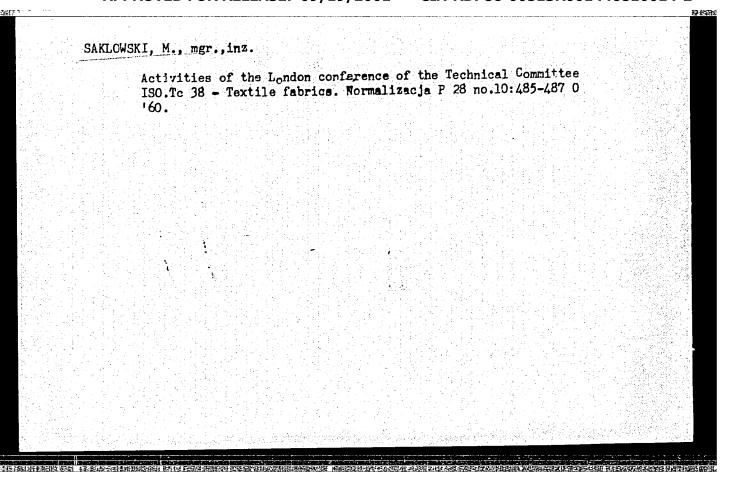
(For Table 2 see next card)

Card 2/3

Engineering and economic...

| Table 2. 1 - Indices 2 - Output per lm² of factory area, tons..... 3 - Productivity of one worker, tons.....
| There are 2 tables.

Silicate glue for pasting microscopic prepara [Ukr.] 12 no.2:90-91 '55.	tions. Bot.zhur. (MLRA 8:10)
l. Glukhivs'kiy pedagogichniy institut (Glue) (Biological specimens)	
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GORYAYEV, M.I.; PUCACHEV, M.G.; TRET'YAKOV, L.I.; POPOV, A.P.; KORNILOVA, G.P.; IBRAYEV, G.Zh.; TUREBEKOV, Sh.S.; SAKWAN, N.E.

Preparation of fodder yeasts from molasses waste of the Dzhambul Alcohol and Vodka Combine. Izv. AN Kazakh. SSR.Ser.khim.nauk 15 no.2:77-82 Ap-Je '65.

(MIRA 18:9)

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SAKNYN, A.V. V-9 USSR/Pharmacology and Toxicology - Toxicology : Ref Zhur - Biol., No 21, 1958, 98626 Abs Jour : Lickens, V.A., Saknyn', A.V. Author Inst : On Hygeinic Evaluation of the General Toxic Action of Title Sulfur Dioxide. : V sb.: Vopr. gigiyeny truda, professional'noy patologii Orig Pub i toksikologii v prom-sti sverdl, obl. Sverdlovsk, 1955, 150-172. : Experiments were conducted on Expophysectomized and intact Abstract frogs. After hypophysectomy, the frogs were sluggish, with sharply decreased notor functions and lowered reaction to external stimuli. Hexamal was introduced subcutaneously in a dosage of 4-40 %/kg and chloral hydrate 40-100 8/kg. Duration and depth of narcosis in hypophysosectomized frogs is greater than in intact frogs. Card 1/2 - 37 -

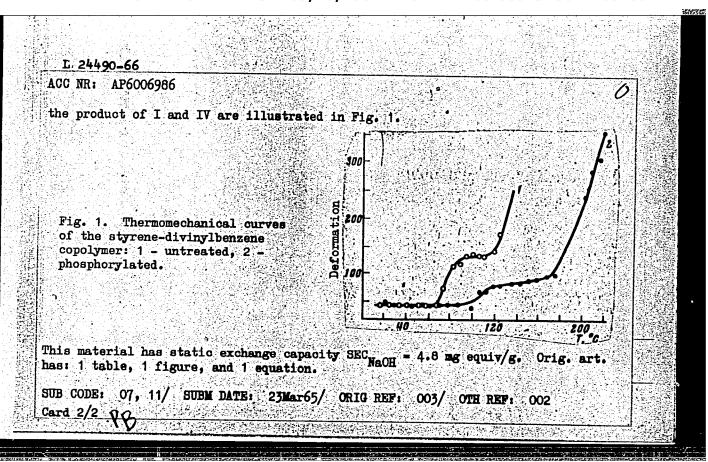
Relationship of the dissolved water content of hydrocarbon fuels to relative humidity at different temperatures. Knim. i tekh.topl. no.11:43-46 N '56. (MLRA 9:11)  1. Nauchno-issledovatel'skiy institut goryuche-smazochnykh materialov. (Hydracarbons) (Fuel) (Humidity)						The second secon
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SABLINA, Z.A.; FOMINA, A.M.; CHURSHUKOV, Ye.S.; SAKODYNSKAYA, T.P.

Evaluation of the performance of sulfur-containing diesel fuels and their components by means of rapid laboratory methods. Khim. i tekh. topl. i masel 8 no.5:57-61 My 163.

(MIRA 16:8)

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AUTI	HORS: Bebikh, G. F.; Sakodynskaya, T. P. $oldsymbol{eta}$
ORG uni	: Moscow State University im. M. V. Lomonosov (Moskovskiy gosudarstvennyy versitet)
PITI	LE: Phosphorylation of polymers
SOUF	RCE: Vysokomolekulyarnyye soyedineniya, v. 8, no. 2, 1966, 343-346
[OP]	IC TAGS: copolymer, phosphorylation, phosphorus compound
(II) gate phos	PRACT: Phosphorylation of styrene divinylbenzene copolymer (I) of polypropylene and polyacrylonitrile (III) with phosphorus pentasulfide (IV) has been investied. This work represents a further development in the study of IV as a new sphorylating agent (see G. F. Bebikh, V. K. Kuskov, V. M. Shatskiy, and G. L.
odl rith it 2 esu	lovchenko, Vysokomolek. soyed. 4, 1677, 1962). Although phosphorylation of I IV at 132C at atmospheric pressure yielded a polymer containing 4.40% of P and 200C in an autoclave8.02% of P, 6.10% S, neither II nor III gave satisfactory alts when treated with IV. The amount of P incorporated into the polymeric
	noture was below the amount obtained with PCl <sub>3</sub> . Thermomechanical properties of UDC: 678.01:54



ACCESSION NR: AP4004702

S/0065/63/000/012/0050/0055

AUTHOR: Englin, B. A.; Rozhkov, I. V.; Tugolukov, V. M.; Sakody\*nskaya,

T. P.

TITLE: Prevention of ice crystal formation in aviation fuels

SOURCE: Khimiya i tekhnologiya topliv i masel, no. 12, 1963, 50-55

TOPIC TAGS: aviation fuel, ice formation, ethylcellosolve, fuel additive, antifreeze

ABSTRACT: A study has been made of the effectiveness of cellosolve (GOST 8313-60) as an antifreeze additive for T-1 and TS-1 jet fuels and B-95/130 aviation gasoline. Previous tests showed that of 100 compounds tested, cellosolve is the most effective. In studying the formation of ice in the fuels in the presence of cellosolve, the following conditions were varied: temperature (down to -60C), moisture lowing conditions were varied: temperature (0-0.3%), ambient relative hucontent (0-0.13%), cellosolve content (0-0.3%), ambient relative hucontent (0-0.13%), the laboratory, in ground storage, and in midity, and storage time in the laboratory, the addition of 0.3% celloflight. Under all the conditions studied, the addition of 0.3% cellosolve completely prevented the formation of ice in the fuels. The

Card 1/2

ACCESSION NR: AP4004702

presence of cellosolve in the fuels had no negative effect on either the physicochemical properties or the performance characteristics of the fuels. Moisture absorption in storage at 30—100% relative humidity was studied in a l-yr test with T-1, TS-1, and B-95/130 fuel samples without cellosolve or containing 0.3% cellosolve. The moisture content of the latter remained nearly the same as that of controls throughout the year. Orig. art. has: 4 tables and 2 figures.

ASSOCIATION: none

SUBMITTED: 00 DATE ACQ: 03Jan64 ENCL: 00

SUB CODE: FL NO REF SOV: 007 OTHER: 007

Card 2/2

"Prepar	rative Columns <sup>10</sup>							
Report in Gas	to be submitted Chromatography,	for the S Houston,	econd In Texas, 2	ternationa 3-26 March	ıl Symposi ı 1964.	um on A	dvances	
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sov/7**156**-58-3-51/52

AUTHORS:

Sakodynskiy, K. I., Babkov, S. I., Zhavoronkov, N. M.

TITLE:

Two-Temperature Method for the Separation of Binary Mixtures (Dvukhtemperaturnyy metod razdeleniya binarnykh smesey)

PERIODICAL:

Nauchnyyo doklady vysshey shkoly, Khimiya i khimicheskaya teknnologiya, 1958, Nr 3, pp. 598-602 (USSR)

ABSTRACT:

In the present paper the most important rules governing the two-temperature method for the separation of binary mixtures are explained. The two-temperature method may be used successfully for the isotopic separation and for the absorption-desorption separation of gases. The conditions for carrying-out effective separations by means of the two-temperature method are given. The degree of elution  $\varphi$  in the two-temperature method is dependent on the temperature difference. An equation was formulated for the approximate determination of the number of theoretical stages necessary to obtain the separation desired. It was found that two separation columns are connected with each other by the two-temperature method and that they reach the same separation effect as can be

Card 1/2

APPROVED FOR RELEASE: 09/19/2001 CIA-RDP86-00513R001446810014-2"

SOV/156-58-3-51/52

Two-Temperature Method for the Separation of Binary Mixtures

reached using a rectifying column with n number stages and the separation coefficient  $\sqrt{\alpha_1/\alpha_2}$ .

There are 2 figures and 5 references, 3 of which are Soviet.

ASSOCIATION: Kafedra tekhnologii svyazannogo azota i shche-

lochey Moskovskogo khimiko-tekhnologicheskogo instituta im. D. I. Mendeleyeva

(Chair for the Technology of Bound Nitrogen and Alkalies at the Moscow Chemical and Technological Institute imeni D. I.

Mendeleyev)

SUBMITTED: October 28, 1957

Card 2/2

5(4)
SOV/20-121-4-30/54
AUTHORS: Sakodynskiy, K. I., Babkov, S. I., Zhavoronkov, N. M.,

Corresponding Member, Academy of Sciences, USSR

TITLE: The Coefficients of the Equilibrium Distribution of Deuterium

in the Isotope Exchange Between Water and Some Thiols (Ko-effitsiyenty ravnovesnogo raspredeleniya deyteriya pri izo-

topnom obmene mezhdu vodoy i nekotorymi tiolami)

PERIODICAL: Doklady Akademii nauk SSSR, 1958, Vol 121, Nr 4, pp 681-684

(USSR)

ABSTRACT: It was interesting experimentally to determine the coefficients

mentioned in the title. This paper investigates the equilibrium of the reactions of deuterium exchange between water

and normal butyl thiol n-C<sub>4</sub>H<sub>9</sub>SH, secondary butyl thiol sec-C<sub>4</sub>H<sub>9</sub>SH, isoamyl thiol iC<sub>5</sub>H<sub>11</sub>SH, normal hexylthiol

 $n-C_6H_{13}SH$ , and thiophenol  $C_6H_5SH$ . The experimental determina-

tion and the calculation of the coefficients  $\alpha$  of the equi-

librium distribution of deuterium are discussed. The isotope equilibrium in the exchange between water and the thiols

Card 1/3 is obtained after 8 - 10 hours at a temperature of 20

SOV/20-121-4-30/54 The Coefficients of the Equilibrium Distribution of Deuterium in the Isotope Exchange Between Water and Some Thiols

> (after 2 hours after the exchange with thiophenol) and after 2 - 4 hours at 80°. In addition to the experiments concerning the direct exchange (between water enriched by deuterium and thiol of a natural deuterium concentration), for each of the investigated types of thiol one experiment concerning the inverse exchange at 20° was carried out. The results of the experimental determination of the coefficient a of the equilibrium distribution (for various temperatures) are given in a table. The corresponding errors are then discussed. Under the discussed conditions of the isotope exchange, only the hydrogen isotopes of the group S-H participate in the reaction. The temperature dependence of  $\alpha$  is shown in a diagram and the corresponding analytic expressions  $lg(\alpha) = f(T)$  for the various thiols are explicitly given. However, the results of this paper and also previous results are not sufficient for the finding of a direct connection between the quantity a and the composition (and the structure) of the radical group . It is only evident, that the influence of the structure and of the composition of the radical group on the value of  $\alpha$  is

Card 2/3

APPROVED FOR RELEASE: 09/19/2001 CIA-RDP86-00513R001446810014-2"

SOV/20-121-4-30/54

The Coefficients of the Equilibrium Distribution of Deuterium in the Isotope Exchange Between Water and Some Thiols

weak. There are 1 figure, 1 table, and 10 references, 5 of

which are Soviet.

ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya. Karpova

(Physical-Chemical Institute imeni L. Ya. Karpov)

SUBMITTED: May 13, 1958

Card 3/3

5(4) 507/156-59-2-9/48

AUTHORS: Sakodynskiy, K. I., Zhavoronkov, N. M.

TITLE: The Rate of Hydrogen Exchange Between Water and Isoamyl Thiol

in an Inert Solvent (Skorost' vodorodnogo obmena mezhdu vodoy

i izoamiltiolom v srede inertnogo restvoritelya)

PERIODICAL: Nauchnyye doklady vysshey shkoly. Khimiya i khimicheskaya

tekhnologiya, 1959. Nr 2, pp 256-259 (USSR)

ABSTRACT: Scarce publication data on the hydrogen exchange between

the S-H- and O-H-groups (Refs 1-11) are mentioned. As the two components mentioned in the title are not soluble in one another, the rate of isotopic exchange depends to a considerable extent on the conditions of mixing of the

two components (Fig 1). Therefore, the reaction in solution

was investigated in an inactive medium, i.e. acetone.

Though acetone enters an exchange reaction with water, this reaction is so slow (Ref 14) in the neutral medium that it was possible to neglect it for the short periods of experi-

menting. The degree of exchange was calculated according to

the following formula:  $F = \left(1 - \frac{x_0 - x_\tau}{x_0 - x_\infty}\right) 100\% (x = initial)$ 

Card 1/2

SOV/156-59-2-9/48
The Rate of Hydrogen Exchange Between Water and Isoamyl Thiol in an Inert Solvent

concentration of deuterium in water,  $x_{\tau} = \text{concentration of}$ D according to the time  $\tau_{\tau}$   $x_{\tau} = \text{concentration of D after the}$ state of equilibrium has been attained). A table gives the experimental results. They indicate that the exchange reaction proceeds in the inert solvent with high velocity and is finished after 0.5 min, and that the isotopic exchange does not cause side reactions. Figure 2 shows the difference of the reaction rate in the case of the mixing of the components and of solving them in an inert solvent. There are 2 figures, 1 table, and 16 references, 9 of which are Soviet.

PRESENTED BY: Kafedra tekhnologii svyazannogo azota i shchelochey Moskovskogo

khimiko-tekhnologicheskogo instituta im. D. I. Mendeleyeva (Chair of the Technology of Bound Nitrogen and Alkalies of the Moscow Institute of Chemical Technology imeni D. I.

Mendeleyev)

SUBMITTED: December 31, 1958

Card 2/2

21(5) sov/64-59-3-9/24

AUTHORS: Zhavoronkov, N. M., Sakodynskiy, K. I.

. TITLE: Industrial Methods for the Production of Heavy Water (Promysh-

lennyye metody polucheniya tyazheloy vcdy)

PERIODICAL: Khimicheskaya promyshlennost', 1959, Nr 3, pp 35 - 48 (USSR)

ABSTRACT: A detailed survey is given on industrial methods of producing

> heavy water, stating data on the technological methods of the whole world, which are projected or have been carried out, or which are being carried out or have been dropped already. The various methods of developing and separating deuterium are theoretically dealt with, as well as the electrolysis, the che-

> mical exchange of isotopes and the rectification of water and hydrogen, the corresponding data are given (Tables 1,2 (print-

ing error)). Among the various factories in the western countries established for the production of heavy water, those works are mentioned which work according to the electrolytic method, the rectification of water and the distillation of

hydrogen, and according to the double-temperature method, the corresponding figures are given, representations in diagrams

Card 1/2 and industrial and commercial data are shown. As far as the

Industrial Methods for the Production of Heavy Water 50V/64-59-3-9/24

method of hydrogen distillation is regarded, it was mentioned that this was the first method to be introduced in USSR industry, deuterium was extracted from electrolytic hydrogen according to the low-temperature rectification (Ref 37). The latter production is the only example given for the production of heavy water in a country other than western and is described in detail (Fig 4, scheme). A table is also given (Table 3) with data on works and firms in western countries which deal or dealt with the production of heavy water. Finally comparisons are made between the various methods of producing heavy water, and the corresponding data are given of some works in the western countries (Tables 4,5). There are 12 figures, 5 tables, and 96 references, 6 of which are Soviet.

Card 2/2

Industrial	production of	heavy water.	Atom taj 2	no.4:73-84 '59.	
그런 그런 경기를 받았다. 그런 글로 보고 보고 있다.					

sov/63-4-1-23/31 5(3)

Sakodynskiy, K.I., Babkov, S.I. AUTHORS:

Mutual Solubility of Water and Some Thiols (Vzaimnaya rastvori-TITLE:

most' vody i nekotorykh tiolov)

Khimicheskaya nauka i promyshlennost', 1959, Vol 4, Nr 1, PERIODICAL:

p 133 (USSR)

The solubility of the following thiols in water and the solubility of water in them has been investigated: normal butyl-ABSTRACT:

thiol, secondary buthylthiol, isoamylthiol, normal hexylthiol and thiophenol. The solubility of the thiols in water is inversely propertional to the molecular weight. At an increase of temperature the solubility of n-butylthiol and thiophenol rises,

but that of isoamylthiol decreases. The solubility of water in

the thiols increases with temperature. Card 1/2

Mutual Solubility of Water and Some Thiols

SOV/63-4-1-23/31

ASSOCIATION: Fiziko-khimicheskiy institut imeni L.Ya. Karpova (Physical-

Chemical Institute imeni L.Ya. Karpov)

SUBMITTED: July 16, 1958

Card 2/2

ZHAVORONKOV, N.M.; BABIKOV, S.I.; ORLOV, V.Yu., kand.khimicheskikh nauk;

SAKODYNSKIY, K.I., kand.khimicheskikh nauk; SEVRYUGOVA, N.N.;

SOKOL'SKIY, V.A.; CHERNYEH, G.N.

Production and uses of stable isotopes. Khim.nauka i prom. 4

no.4:487-498 '59. (MIRA 13:8)

(Isotopes--Industrial applications)

21(1) AUTHOR:

Sakodynskiy, K. I.

SOV/89-6-1-2/33

TITLE:

Industrial Production of Heavy Water (Promyshlennoye

proizvodstvo tyazheloy vody)

PERIODICAL:

Atomnaya energiya, 1959, Vol 6, Nr 1, pp 14 - 20 (USSR)

ABSTRACT:

This is an abstract from 80 articles by Western authors dealing with the following subjects concerning heavy water:

- 1. Probable heavy water consumption in various countries (Great Britain, France, India, Australia, Italy, Switzerland, Sweden, Norway, and Western Germany).
- 2. The manufacturing plants at present existing in various countries are described in short.
- 3. The plans for the extension of various manufacturing plants in various countries are mentioned.

There are 2 figures, 1 table, and 80 references, 3 of which are Soviet.

Card 1/2

Industrial F	roduction	of Heavy Water	SOV/89-6-1-2/33
SUBMITTED:	May 4,	1958	
Card 2/2			

0.0000 77230 SOV/89-8-1-24/29

AUTHOR: Zhavoronkov, N. M., and Sakodynsky, K. I.

TITLE: Scientific and Technical News. At the Institute of Physi-

cal Methods of Separation (German Democratic Republic)

PERIODICAL: Atomnaya energiya, 1960, Nr 1, pp 81-82 (USSR)

ABSTRACT: In September of 1959, through an invitation by the German

Academy of Sciences in Berlin, the authors of this article visited the Institute of Physical Methods (headed by J. Muhlenfordt) in the city of Leipzig. It is the center, in the German Democratic Republic, of experimental work on the stable isotopes. The Institute was organized in 1955. Its basic requirement is the expansion of research and cooperation on the use of stable isotopes among the scientific and other related organizations. The Institute

does a great deal of work on the application of the stable isotopes to chemistry, biology, medicine, geology, etc.; it also developes the methods of working with the

Card 1/2 stable isotopes and works on the theory of separation

Scientific and Technical News. At the Institute of Physical Methods of Separation (German Democratic Republic)

77230 SOV/89-8-1-24/29

processes. There are six departments in the Institute: (1) the experimental separation through rectification, headed by E. Kröll; (2) the experimental separation through the chemical exchange, headed by K. Wetzel; (3) the theory of separation methods, headed by G. Fogt; (4) the theoretical, headed by G. Voigt; (5) the analytic, headed by G. Birkenfeld; and (6) the application of stable isotopes, headed by H. Hübner. All the above departments are involved in production of stable isotope concentrates of hydrogen, boron, carbon, nitrogen, and oxygen. Construction is scheduled of the new powerful units for an expanded production of stable isotopes and the concentrates BO by a method of the exchange distillation between BF3 and its groupings with anisole. Besides

fulfilling its own requirements, the Institute performs the isotope analysis for all related organizations in the German Democratic Republic.

Card 2/2

S/074/60/029/009/002/002 B013/B064

AUTHORS:

Sakodynskiy, K. I., Zhavoronkov, N. M.

TITLE:

Two-temperature Methods of Producing Heavy Water 19

PERIODICAL:

Uspekhi khimii, 1960, Vol. 29, No. 9, pp. 1112-1135

TEXT: The present survey deals with methods of producing heavy water, the main laws of two-temperature methods, and the systems of the substances used. Methods of producing heavy water are given in a number of surveys (Refs. 12-32) and in a monograph (Ref. 33). In principle, all methods of isotope separation are suited for the concentration of heavy water. The following methods have found industrial application: electrolysis of water combined with the isotopic exchange between water and hydrogen, rectification of water, low-temperature rectification of hydrogen, and two-temperature exchange between water and hydrogen sulfide. The two last-mentioned methods have gained ever-increasing importance during the past ten years (Ref. 34). The principal methods given are usually combined with one another according to the operating conditions. All production processes of D<sub>2</sub>O can be divided into independent methods, and into methods depending Card 1/3

Two-temperature Methods of Producing Heavy Water

S/074/60/029/009/002/002 B013/B064

on the production of other substances. The distillation of water and the two-temperature exchange between water and hydrogen sulfide belong to the former. The others comprise all the other methods whose capacity is limited by the amount of hydrogen production for works of ammonia synthesis. Table 1 gives some main indices of the different methods of synthesis. The table indicates that the two-temperature exchange and the low-temperature rectification of water are the most favorable methods from the economic point of view. These two methods are applied in several countries on an industrial scale (Refs. 11, 43-47). A list of the plants in operation and of the biggest projects of heavy-water production is published in Ref. 30. Refs. 34, 48, 49 give data on the production of heavy water in several countries. The main laws of the two-temperature method are described (Figs. !-7 and Refs. 12, 19-25, 30, 31, 35, 50-62). This method can be applied in two ways: 1) by cascade-like arrangement of individual pairs of multi-stage apparatus; one of the apparatus operates at a low, the other at a high temperature (Fig. 1); 2) by linking two multi-stage columns one of which operates at a high, the other at a low temperature (Fig. 2). Requirements to be met in selecting the system of substances are given. Table 2 gives a comparison of some systems of substances. It Card 2/3

Two-temperature Methods of Producing Heavy Water

S/074/60/029/009/002/002 B013/B064

shows that the highest deuterium yield is obtained when applying the systems water - hydrogen, water - hydrogen halide, or ammonia - hydrogen. The following systems are described: water - hydrogen: Tables 3, 4, Figs. 8, 9 (Refs. 12, 15, 19, 24, 36, 45, 57, 66-76); ammonia - hydrogen: Table 5; Figs. 10-12 (Refs. 33, 77-84); water - hydrogen sulfide: Tables 6, 7, Figs. 13, 14 (Refs. 12, 15, 19, 35, 40, 45, 48, 51, 57, 73, 79, 85-127); other systems: Tables 8-10, Fig. 15 (Refs. 9, 18, 35, 62-64, 128-132). I. V. Kurchatov, A. M. Rozen, Ya. M. Varshavskiy, and S. E. Vaysberg are mentioned. There are 15 figures, 10 tables, and 132 references: 16 Soviet.

ASSOCIATION:

Fiziko-khimicheskiy institut im. L. Ya. Karpova (Physicochemical Institute imeni L. Ya. Karpov)

Card 3/3

POTAPOV, S.P.; SAKODYNSKIY, K.I.; BORISOVSKAYA, M.A., red.; VIASOVA, N.A., tekhn. red.

[Stable isotopes around us] Stabil'nye izotopy vokrug nas. Moskva, Gos. izd-vo lit-ry v oblasti atomnoi nauki i tekhn., 1961. 67 p.

(Isotopes)

(Isotopes)

MALAFEYEV, N.A.; MALYUSOV, V.A.; UMNIK, N.N.; SAKODYNSKIY, K.I.; ZHAVORNOKOV, N.M. Prinimali Uchastiye: PODGORNAYA, I.V.; ABRAMOVA, V.F.; BARANOVA, V.I.

Determination of the fractionation factors of binary mixtures tetrachloroalkanes during vaporization in a high vacuum. Enim. prom. no.3:196-198 Mr '61. (MIRA 14:3)

(Paraffins) (Distillation, Fractional)

Use of high Khim.prom.	-frequency anal no.12:868-870	ysis in the study of D 161. (Absorption)	absorption kinetics. (MIRA 15:1)	

S/081/62/000/017/052/102 B158/B186

AUTHORS:

Zavoronkov, N. M., Sakodynskij, K. I.

TITLE:

Industrial manufacture of heavy water

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 17, 1962, 353, abstract

17K8 (Kernenergie, v. 4, no. 12, 1961, 893-904 [Ger.])

TEXT: A review is presented. Methods of producing heavy water and information on plants and units for heavy water production constructed, in course of construction, and planned are given. Data are presented on the dynamics of variation in the productive capacity for heavy water, and data on the costs of producing heavy water by various methods are compared. 63 references. [Abstracter's note: Complete translation.]

Card 1/1

SAKODYNSKIY, K.I.; KHOKHLOVA, L.A.

Effect of the medium on the rate of deuterium exchange between water and isoamylothiol. Zhur. VKhO 6 no.6:705-706 '61. (MIRA 14:12)

1. Fiziko-khimicheskiy institut imeni L.Ya.Karpova. (Water) (Thiols) (Deuterium)

khim. 16 no.3:375-	y mixtures of tetrachloroalkanes. Z -376 My-Je '61. Scientific Research Physico-Chemical		
1. L. Ya. Karpov s Moscow.	(Paraffins)		
	보는 말로 마르는 것 같은 말로 가게 말을 수 되었다.		
	이라면 그렇게 한 테르크 프랑스 함께 하다라면 하시고를 보고 하기를 보고 있다.		

	KISELEV, A.V.; SAKODYNSKIY, K.I.; SHCHERBAKOVA, K.D.  Fourth International Symposium on Gas Chromatography in Hamburg. Neftekhimiia 2 no.5:804-814 S-0 '62. (MIRA 16:1) (Hamburg-Gas chromatography-Congresses)			www1-4-4-4	

SAKODYNSKIY, K.I.; BABKOV, S.I.; ZHAVORONKOV, N.M. (Moscow)

Isotopic hydrogen exchange between water and thiols. Zhur.fiz.khim.
36 no.10:2169-2175 0 '62. (MIRA 17:4)

1. Fiziko-khimicheskiy institut imeni Karpova, Moskva.

RAKOVSKIY, V.S.; SAKLINSKIY, V.V.; FILIMONOV, V.C., inzh., retsenzent;
MARTENG, S.L., Inzh., red.; GORDEYEVA, L.P., tekhn. red.

[Powder metallurgy in the machinery industry] Poroshkovaia
metallurgiia v mashinostroenii; spravochnik. 2 izd., ispr.
i dop. Moskva, Mashgiz, 1963. 101 p. (MIRA 16:8)

(Powder metallurgy—Handbooks, manuals, etc.)

Automatic sampling in preparative chromatography. Khim topl.i masel 8 no.11:49-52 N '63.	i tekh. (MIRA 16:12)
1. Fiziko-khimicheskiy institut im. L.Ya.Karpova.	

BRAZ, G.I.; MYASNIKOVA, G.V.; YAKUBOVICH, A.Ya.; BAZOV, V.P.;
SAKODYNSKIY, K.I.

Simultaneous trimerization of acetonitrile and trichloroacetonitrile.
Zhur.ob.khim. 33 no.6:1939-1941 Je '63. (MIRA 16:7)

1. Fiziko-khimicheskiy institut imeni L.Ya.Karpova.
(Acetonitrile) (Polymerization)

Fourth Symposium on no.9:76-77 S '63.	Gaseous Chromatography.	(MIRA 16:9

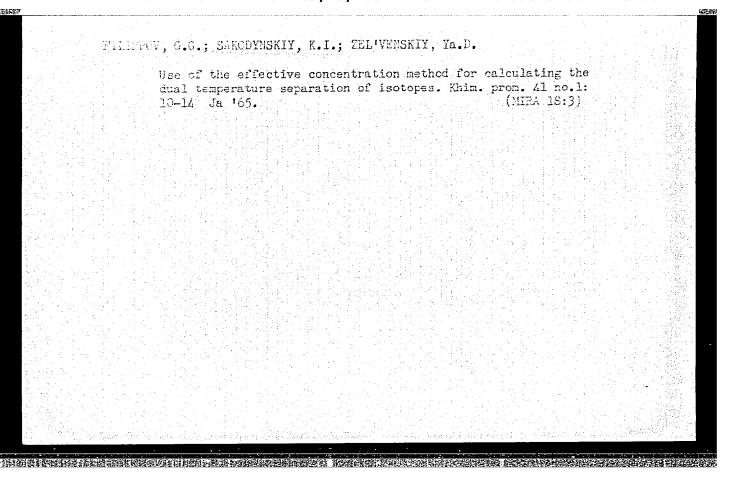
Busic trends in the development of the methods of separation of stable isotopes. Zhur. prikl. khim. 36 no.12:2564-2579 D'63. (MIRA 17:2)
1. Fiziko-khimicheskiy institut imeni Karpova.

	W, R.V., OMODYHOKIY, K.I., red.		
	[Molecular chromategraphy] Moleku Moskva, Nauka, 1964. 161 p.	liarnaia khromatografija. (MIRA 17:11)	
	1. Chlen-korrespondent AN SSSR (f	or Chuntov).	
		그리아 그 10 1000 등 경기 교육 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
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S/0089/64/017/001/0070/0071 ACCESSION NR: AP4042268 AUTHOR: Volkov, S. A.; Sakody\*nskiy, K. I. TITLE: Gas chromatographic separation of labeled compounds SOURCE: Atomnaya energiya, v. 17, no. 1, 1964, 70-71 TOPIC TAGS: gas chromatograph, column chromatograph, preparative chromatography, chromatographic separation, labeled compound separation, deuterated methylethoxysilane separation, siloxane rubber preparation ABSTRACT: A preparative gas chromatographic technique, heretofore virtually unused for the separation of labeled compounds, has been applied to the separation of deuterium-labeled trimethylethoxysilane(I)

and dimethyldiethoxysilane(II) out of the mixture produced in the reaction of tetraethoxysilane with deuterium bromide. I and II are used in the preparation of deuterated siloxane rubber. A detailed description is given of a semiautomatic column chromatograph which permits manual introduction of samples and automatic trapping of separated fractions. The operating procedure of the chromatograph

total of 8 runs at a col	ral grams of I and II were umn temperature of about 1 t possible to separate con i.e., even higher boiling	100C. At that npounds with boiling
ASSOCIATION: \ none		
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SUB CODE: GC	NO REF SOV: 000	OTHER: 001



SAKON TIKOV N.T.

PHASE I BOOK EXPLOITATION

SOV/4325

USSR. Gosudarstvennyy komitet po radioveshchaniyu i televideniyu

Radio i televideniye v SSSR (Radio and Television in the USSR) Moscow, 1960. 164 p. 4,000 copies printed.

Editorial Board: S.V. Kaftanov, N.P. Kartsov, N.I. Sakontikov, M.S. Gleyzer, and P.S. Mozharovskiy; Tech. Ed.: Ya. Dubson.

PURPOSE: This book is intended for the general reader.

COVERAGE: The book gives a description of the main features of Soviet radio and television. Information is given on radio and television programs transmitted from central and local stations in the USSR, on radio and television publications, on the volume of broadcasting, and on broadcasts to non-Soviet listeners. The activity of the Vsesoyuznyy nauchno-issledovatel'skiy institut zvukozapisi (All-Union Scientific Research Institute of Sound Recording) and of the Gosudarstvennyy dom radioveshchaniya i zvukozapisi (State House of Broadcasting and Sound Recording) is described. No personalities are mentioned. There are no references.

Card 1/9

Radio and Television in the USSR SOV/4325 TABLE OF CONTENTS:	
Foreword [S. Kaftanov, Chairman, State Committee for Broadcasting and Televisi Council of Ministers, USSR]	on, 3
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Organization and Structure of Soviet Radio and Television	19
Programs Broadcasted From the Center to the Population of the USSR  Radio Broadcasting  Central radio broadcasts  "Latest news"	22 27 27
Card 2/9	with Pierr

	Notes on Japanese television technology. Tekh.kino i telev. 4 no.8:93-95 Ag '60. (MIRA 13:8) (Japan-Television)	
	환경 경기 내용을 가득하는 일 말을 하는데 되는데 말을 하는데 하는데 없다.	
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	물이 된 일본 사람이 그림 만들는 그를 보는 아니라 살고 있었다. 그리고 있다면	
	'문화생활 '발송화울'은 사람이 없었습니 회사를 잃어 하고만 그리고 되면 하고 되었다.	
	왕일 및 본역로 중인 대답입상 등시간로 살이 하면 가능으로 이 것이 걸었다.	
	그를 하고 있는데, 전문이 보다 하면 이 사람들은 하고 있었다. 그리고 있는데 그 없었다. 나는	
	이는 계획이다고 하는 것들이 많은 사람들은 그렇게 하지만 하면 하는 모르는 모르겠다.	
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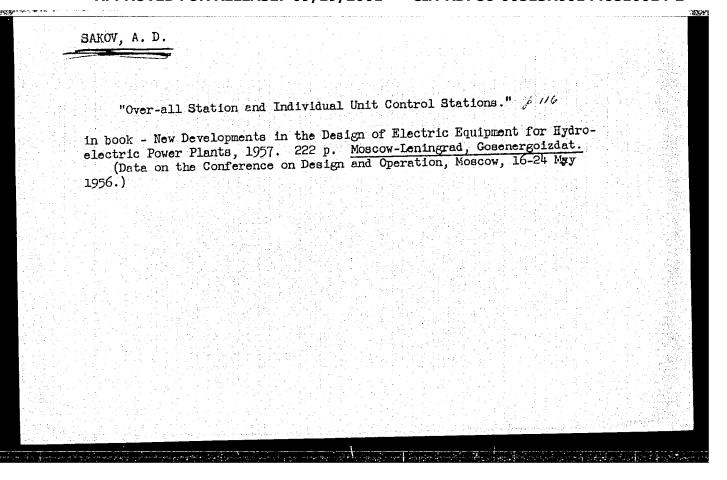
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OSTROVITYANOV, K.V., akademik; GATOVSKIY, L.M.; KUZ'MINOV, I.I., doktor ekon. nauk; Prinayali uchastiye: STAROVSKIY, V.N.; SAKOV, M.P.; BACHURIN, A.V.; ZASLAVSKAYA, T.I.; EOGOMOLOV, O.T.; RYMALOV, V.V.; RABINOVICH, M., red.; MUKHIN, Yu., tekhn. red.

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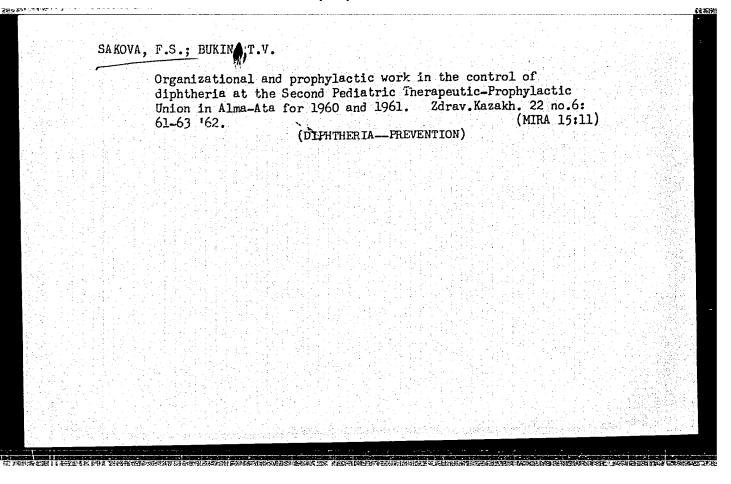
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# SAKOVA, A.A.; SHOTOV, A.P., kand.fiziko-matem.nauk

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L 130L7-65 EWT(m)/EWP(w)/EPF(n)-2/EWA(d)/EWP(t)/EWP(b) Pu-L/Pad AS(mp)-2/AEDC(a)/AFWL/SSD/ASD(m)-3/ASD(a)-5/ESD(t) JW/JD/HW/JC/MLK ASD(2)-2/ \$/0000/64/000/000/0150/0154 ACCESSION NR: AT4046834 AUTHOR: Popov, L. Ye.; Sukhovarov, V. F.; Panova, L. H.; Sakova, H. P. в TITLE: Effect of atomic defect relaxation on diffusion transformation in Nialloys SOURCE: AN SSSR. Nauchny\*y sovet po probleme zharoprochny\*kh splavov. Issledovaniya staley i splavov (Studies on steels and alloys). Hoscow, izd-vo Nauka, 1964 150-154 TOPIC TAGS: diffusion, nickel, molybdenum, nickel molybdenum alloy, activation energy, migration energy, diffusion transformation, atomic defect, atomic defect relexation ABSTRACT: During stepwise tempering or heating at a constant rate, cold-worked nickel reveals two stages of atomic defect relaxation at temperatures higher than room temperature, evidenced by the recovery of electrical resistance and density and the emission of absorbed energy. It was therefore considered desirable to investigate the effect of atomic defect relaxation on transformations in cold-worked nickel alloys with a large difference between the atomic radii of the alloy components. In the present paper, an alloy was selected with 10 at % Ho. The difference between the atomic radii in this alloy could affect its diffusion in Card 1/3

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YENIN, V.T., kand.tekhn.nauk; SAKOVICH, A.A., kand.tekhn.nauk;
FILIMONOV, A.N., inzh., (Leningrad).

Prospective use of d.c. electric power transmission in the Soviet
Union. Electrichestvo no.11:88-92 N '57. (MIRA 10:10)

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elektrotekhnicheskiy institut im. Lenina (for Sakovich).

(Electric power distribution)

4.2. (14 H. 14 J. Sakovich, A.A., Candidate of Technical Sciences (All-Union Electro-technical Institute VEI.) The investigation of materials for mercury-arc rectifiers. AUTHOR: (Issledovaniye materialov dlya rtutnykh vypryamiteley.) PERIODICAL: "Vestnik Elektropromyshlennosti" (Journal of the Electrical Industry.) 1957, Vol. 28, No. 5, pp. 39 - 46, (U.S.S.R.) This article examines questions relating to materials used for the electrodes (anodes and grids), for internal components (fixing parts and screens) and for the vacuum-tight casing i.e. to special steels and graphite. The main characteristics ABSTRACT: of materials for the internal parts of a valve are the following: stability to ionic bombardment; the magnitude of electronic and ionic emission; gas evolution on heating in The stability to bombardment by ions of mercury was investivacuum; and diffusion of hydrogen. gated in an instrument resembling a model of a mercury valve. The specimen which is in the form of a rounded disc, is submitted to intense bombardment by mercury ions for which mitted to intense bombardment by mercury ions for which purpose a high negative potential is applied to it. The materials examined were several grades of graphite and of steel, molybdenum, tungsten, nickel and titanium, and their performance is reported in terms of frequency of backfire at performance is reported in terms of frequency of backfire at an analysis of the support of 20 kV with a current of 0.5 mA; pulverisation with an ion energy of 20 keV and gas evolution on heating in vacuum to a temperature of up to 850 °C.

The investigation of materials for mercury-arc rectifiers. (Cont.)

Emission was investigated on a cylindrical graphite specimen 3 mm in diameter and 25 mm long fixed in terminals. The specimen could be heated to a high temperature by passage of current. The receiving electrode was a molybdenum cylinder, the whole equipment was placed in a high vacuum. Investigations of graphite specimens of different compositions showed that of graphite specimens of different compositions showed that of backfires, grades of graphite containing admixtures of some of backfires, grades of graphite containing admixtures of some fractions of a percent gave a great increase in frequency of backfire with increase in temperature. Tests were made on samples of graphite containing various contaminants. Oxides of some elements (e.g. Sio and Aloo) give much greater emission some elements (e.g. Sio and Aloo) give much greater emission alkali and alkali earth metals give particularly great emission.

Secondary emission of electrons under the influence of ionic combardment was investigated in a special instrument. The investigations showed that the secondary emission current of an undegassed specimen is large at first, then rapidly falls and reaches a steady value after ten or fifteen minutes. The coefficient of secondary emission increases with increasing voltage reaching values greater than unity in the equipment used voltages of 25 to 30 kV. The secondary emission coefficient at voltages with increase in the intensity of the electric field increases with increase in the intensity of the secondary and the surface of the material investigated. The secondary emission coefficient of polished specimens is lower than that

the investigation of materials for mercury are rectifiers, (Cont.)

of rough ones.

The evolution of gas from specimens on heating in a vacuum was investigated in a special equipment. The results for various kinds of steel are presented in the form of a graph. It may be concluded that when steel is degassed in vacuum and exposed to the atmosphere it does not quickly absorb gas. Gas evolution diminishes sharply with reduction in the temperature because of reduced diffusion of gas through the metal. In graphite gas evolution increases after an interruption in the heating. This is associated with absorption of gas by the surface layers of graphite on cooling.

In addition to measuring the quantity of gas evolved the composition was determined with a magnetic mass spectrometer. In investigating gas evolution from steel it was found that the most undesirable components are oxygen, carbon dioxide and heavy hydro-carbons. Water, carbon monoxide, methane and ammonia were less dangerous. Nitrogen is harmless in small concentrations, hydrogen is not dangerous even at high concentrations because it facilitates the removal of oxides and contaminants. However, a high concentration of hydrogen causes constriction of discharge and overheating of parts and also

hinders ignition.

The diffusion of hydrogen was investigated. It is well-known that deterioration of vacuum takes place by diffusion of hydrogen from the cooling water through the metal walls.

The investigation of materials for mercury-arc rectifiers. (Cont.)

Diffusion was investigated on a hollow cylindrical steel specimen, 100 mm in diameter, 250 mm high with a wall thickness of 4 mm made of the material under investigation. The experiments showed that the quantity of hydrogen diffusing was considerably greater than that of gas leaking-in which was checked separately. The use of anti-corrosion solutions and thin electro-plating did not stop diffusion, special lacquers and enamels were not sufficiently reliable and in any case could not withstand the high temperatures used in degassing. Diffusion was not observed through stainless steel.

The selection of materials and the investigation of the technology of manufacture is next considered. Recommendations are made about specifications for graphite and steel. Information is given about the precautions that must be taken in making parts from graphite in order to avoid contamination and of investigations that have been made into the process of heating graphite in a vacuum. The appropriate treatment for steel is also described. Data is given about the reduction in the frequency of backfiring that results from heat treatment of the graphite. It is recommended that after the graphite has been heated the oven should be filled with a special unnamed gas which greatly speeds up the subsequent degassing of the graphite in vacuum. 8 figures, 4 literature references (Russian).

CIA-RDP86-00513R001446810014-2"

APPROVED FOR RELEASE: 09/19/2001

110-9-2/23 AUTHOR: Sakovich, A.A., Candidate of Technical Sciences, Khudyakov, V.V., Candidate of Technical Sciences, Lazarev, N.S. and

Barakayev, Kh.F., Engineers.

TITLE: An Investigation into the Possibility of Autonomous Supply of the Auxiliary Power Requirements of High-voltage Mercury

Valves. (Issledovaniye vozmozhnostey avtonomnogo pitaniya

sobstvennykh nuzhd vysokovol'tnykh rtutnykh ventiley)

PERIODICAL: Vestnik Elektropromyshlennosti, 1957, Vol.28, No.9, pp. 3 - 8 (USSR).

ABSTRACT: The rectifier/inverter sub-stations of high-voltage d.c. transmission systems use bridge-connected rectifiers whose cathodes may be at very high voltages to ground. The mercury valves require some 1 - 3 kW of auxiliary power, at cathode potential, for ignition excitation, anode heating, and electrode control. It is very difficult to supply the power at the necessary voltage, and special isolating transformers are used which often require to be connected in cascade. It has recently been proposed to tap the power from the damping circuit between the valve anode and cathode. This circuit comprises a series capacitance and resistance used as a potential divider and is card 1/4 usually an essential part of the converter. Control signals are

An Investigation into the Possibility of Autonomous Supply of the Auxiliary Power Requirements of High-voltage Mercury Valves.

transmitted by a modulated light ray which acts on a photo-cell operating at valve potential. The method obviates an isolating transformer and is simpler and cheaper. The principles of tapping power from the damping circuit are then explained. Fig.1 shows the valve bridge of a rectifier/inverter sub-station for the d.c. Stalingrad-Donbas system. The rated voltage of the bridge is 100 kV and the transmitted current 900 A. The three-phase output of the transformer is at 83 kV. The principal operating conditions of a sub-station are considered and an expression is written for the voltage in each case. It is shown that the inverse-voltage contains only the fundamental frequency and multiples of three. The relationship between the harmonic content of the voltage and the fixing angle is shown in Fig. 2 and it is concluded that a filter must be provided in order that power may be tapped from the damping circuit. corresponding circuit is shown in Fig. 3a. The only additional equipment required is a transformer with an insulation level of  $10~{
m k} {
m ar v}$  . The procedure for calculating the maximum power from a tapping is described and the simplifying assumptions underlying the calculation are stated. A vector diagram for the equivalent Card 2/4circuit (Fig. 3b) is used to construct graphs of the active

An Investigation into the Possibility of Autonomous Supply of the 110-9-2/23 Auxiliary Power Requirements of High-voltage Mercury Valves.

(Fig. 4) and reactive (Fig. 5) power as functions of the circuit parameters. Fig. 6 gives the reactive power and the loss in the choke coil as functions of the capacitance for various alues of the capacitance in the damping circuit. Hence, the useful power from the tapping is determined and it is shown that some increase in the capacitance of the damping circuit extends the useful range of power tapped. The power calculations were verified experimentally on a model of the circuit. The damping and tapping circuits were connected in parallel with a thyratron model of a power system sub-station. Voltage oscillograms were taken with firing angle values and transmitted current corresponding to the main operating conditions. The results were worked out on a scale corresponding to the Stalingrad-Donbas scheme and showed that for firing angles close to 0 or 150 (which correspond to normal transmission conditions) the voltage wave-form was satisfactory. For angles near 90 the voltage waveform was very distorted. This was because of insufficientlyclose tuning of the tapping circuit and non-linearity of the inductance of the choke. If the choke is linear the voltage distortion is much less. For firing angles close to 0 and 150°

Card3/4 the tapped voltages and power are in good agreement with the

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An Investigation into the Possibility of Autonomous Supply of the Auxiliary Power Requirements of High-voltage Mercury Valves. 110-9-2/23

calculated values. The proposed circuit has a number of advantages compared with supplies from isolating transformers. It requires only a capacitor, a transformer and a choke, with an insulation level of 10 kV, which are much cheaper than an isolating transformer with an insulation level of 400 kV. A disadvantage of the system is the need to raise the capacitance of the damping circuit. The circuit can be used for all transmission system valves except shunting valves. It can also be used successfully for low-voltage mercury-arc rectifiers in industry and traction to deliver power at voltages of 0.8 - 15 kV. It is best suited to sealed-off valves and has limitations when applied There are 6 figures and 2 Slavic references.

ASSOCIATION: All-Union Electrotechnical Institute (VEI)

April 12, 1957.

AVAILABLE: Library of Congress. Card 4/4

110-9-3/23

Butayev, F.I., Klimov, N.S., Sakovich, A.A. and Stepanov, N.P., Candidates of Technical Sciences. AUTHOR:

11/11/11

High-voltage Rectifiers/Inverters for Direct Current Power TITIE:

Transmission. (Vysokovol'tnyye preobrazovateli dlya pere-

dachi energii postoyannym tokom)

PERIODICAL: Vestnik Elektropromyshlennosti, 1957, Vol. 28, No. 9, pp. 8 - 14 (USSR)

ABSTRACT: Brief reviews of the main stages of development of highpower, high-voltage mercury valves at the All-Union Thermotechnical Institute. The first stage concerns the Kashira-Moscow experimental transmission line. The second stage includes high-voltage valves of intermediate power and the production of single experimental installations. The third stage concerns high-power, high-voltage valves suitable for practical highvoltage d.c. transmission systems. High-voltage rectifiers have been under development at the All-Union Thermo-technical Institute since 1935. A number of the principles then evolved, including single-anode construction, a sectionalised anode system, and oil-cooling, are still used. German experience with d.c. transmission was notable for the fruitful work, directed by G. Dobke, on the development of mercury-arc rectifiers for 150 A Cardl/5 max. and 120 kV max. Joint work with the Scientific Research

High-voltage Rectifiers/Inverters for Direct Current Power Transmission.

Institute for Direct Current (NIIPT) using the Kashira-Moscow experimental transmission line, showed that equipment for d.c. power transmission should be considered as a complex whole. In this experimental system, great difficulties were encountered as a result of instability of valve characteristics and a number of specific properties of high-power, high-voltage, rectifier circuits. The valves are subject to back-fire, break-down, loss of control action of the grids, los of excitation, flashover of anode insulators, and over-voltages on the auxiliary electrodes. There were also difficulties arising from system disturbances such as over-voltages, high-frequency oscillations and current surges. For example on the Stalingrad-Donbas line, where the normal voltage on a valve should be 65 kV, calculations show that during over-voltages it may reach 250 kV. Considerable difficulties arose as a result of free oscillations which appeared in the circuit during ignition and extinction of a valve. Highfrequency oscillations, which caused severe radio-interference and damage to the anodes were suppressed by special reactors. Iowfrequency oscillations were more difficult to suppress, and damper circuits consisting of capacitance and resistance are being used for this purpose. At first, the Kashira-Moscow line operated with one bridge having three series valves in each arm

rd2/5

High-voltage Rectifiers/Inverters for Direct Current Power Transmission.

and later with two bridge circuits connected in series, one having one valve per arm and the other two. In designing the Stalingrad-Donbas line, it is proposed to use a sub-station circuit consisting of eight bridge circuits in series, each for a voltage of 100 kV and a current of 900 A. In all, there will be 192 valves working as converters and 32 "shunting" valves. At full load, each alve will handle a power of 7.5 MW. A serious problem was the control of individual circuits from the auxiliary supply panel which was solved under the guidance of Candidate of Technical Sciences I.L. Rubinshteyn. Control is effected by means of a light ray acting on photo-resistance cells which operate relays. For development work, the All-Union Electro-technical Institute (VEI) has constructed equipment for making static tests at up to 200 kV r.m.s., surge generators and a high-power equivalent for testing valves under rated conditions. A large test installation with a transformer group of 120 MW has been constructed at one sub-station. Questions of valve design are then discussed. A sectional drawing of a high-voltage valve is shown in Fig.1. The distinctive feature of high-voltage valves is the anode system and different methods of sectionalising the anodes are shown in Fig. 2. It is usually advisable to extend anode sectionalisation rd3/5 to the discharge space, and different types of screens are shown

110-9-3/23 High-voltage Rectifiers/Inverters for Direct Current Power Transmission.

in Fig. 3. Other anode constructions are shown in Fig. 4; Fig. 4zh shows an anode design developed by V.D. Andreyev. Sevenl types of valves developed in the All-Union Electro-technical Institute are illustrated in Fig. 5 (photos). The main relationships of voltage division between the intermediate electrodes of the anode assembly were established during the investigations of high-voltage valves. When the valve works as a rectifier, the negative voltage drop occurs mainly in the gap between the anode and the first anode insert. When working as an inverter the positive blocking voltage lies very largely in the space between the insert furthest from the anode and the upper grid. Development of rectifiers proceeded alongside theoretical and physical investigations, of valve strength, current distribution in the anode, and voltage distribution in the anode assembly, ignition effects, effects during failure of operation and during heavy overloads. In the investigation of physical effects, probe methods of vapour density measurement were very useful. Graphs of the probability of back-fire for the valves BP-1 and BP-3 as functions of cooling temperature are shown in Fig. 6. Laboratory results have been confirmed by data of 12 000 hours tests on valve type BP-3 in the experimental transmission line. Card4/5 Valves type BP-4/2 and BP-9/3 are intended for the Stalingrad-

110-9-3/23

High-Voltage Rectifiers/Inverters for Direct Current Power Transmission.

Donbas transmission line and have passed a wide variety of laboratory tests and the results of the first high-power tests confirm the satisfactory operation of the valves in approximately their normal working conditions. In the light of the experience gained, development is likely to comprise: valves of greater power for higher voltages and currents, which will be combined with improvement in manufacture to improve reliability; valve designs which will ensure great vacuum-tightness, enabling the manufacture of pumpless valves or, if necessary, to provide internal devices to maintain and restore vacuum. Two new types of pumpless valve are shown in Fig. 7, one for 150 A max. and the other for 900 A max. with reverse voltages of the oder of 100 - 120 kV max. There are 7 figures and 5 Slavic references.

ASSOCIATION: All-Union Electrotechnical Institute (VEI)

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